

Partie 2:

1) Equilibre d'un élément de vapeur

$$[-P(z+dz) + P(z)] L(\delta-y) + \rho_L g L(\delta-y) dz - \mu_L \frac{du}{dy} dz L = 0$$

$$\Rightarrow \left(-\frac{dP}{dz} + \rho_L g \right) (\delta-y) = \mu_L \frac{du}{dy}$$

Dans la vapeur $\frac{dP}{dz} = \rho_v g$

$$\Rightarrow \mu_L \frac{du}{dy} = (\rho_L - \rho_v) g (\delta-y)$$

2) Conditions aux limites $\frac{du}{dy} = 0$ en $y=\delta$ et $u=0$ en $y=0$

d'où $u(y) = \frac{(\rho_L - \rho_v) g}{\mu_L} y (\delta y - \frac{y^2}{2})$

$$3) m(z) = \rho_L L \int_0^z u(y, z) dy = \rho_L L \frac{(\rho_L - \rho_v) g}{\mu_L} \left[\frac{\delta y^2}{2} - \frac{y^3}{6} \right]_0^z = \frac{\rho_L (\rho_L - \rho_v) g L \delta^3}{\mu_L} \frac{z^3}{3}$$

4) bilan thermique

$$dq = h_L \frac{dT}{dy} \quad \text{densité de flux.} \quad \text{Puissance } dw = h_L \frac{dT}{dy} L dy$$

$$5) dw = h_L \frac{(T_{sat} - T_c)}{\delta} L dy = h_{Lv} dm = h_{Lv} \frac{\rho_L (\rho_L - \rho_v) g L \delta^2}{\mu_L} \frac{dz}{dy}$$

$$\Rightarrow \delta = \left[\frac{4 \mu_L \rho_L (T_{sat} - T_c)}{\rho_L (\rho_L - \rho_v) g h_{Lv}} z \right]^{1/4}$$

$$6) dm = \frac{\rho_L (T_{sat} - T_c)}{h_{Lv}} L dz \times \left[\frac{\rho_L (\rho_L - \rho_v) g h_{Lv}}{4 \mu_L \rho_L (T_{sat} - T_c) z} \right]^{1/4}$$

$$m(z) = \left[\frac{h_L (T_{sat} - T_c)}{h_{Lv}} \right]^{3/4} L \left[\frac{\rho_L (\rho_L - \rho_v) g}{4 \mu_L} \right]^{1/4} z^{3/4} \times \frac{4}{3}$$

$$m(z) = \left[4 \frac{h_L (T_{sat} - T_c)}{h_{Lv}} \right]^{3/4} \left[\frac{\rho_L (\rho_L - \rho_v) g}{\mu_L} \right]^{1/4} \frac{L}{3} z^{3/4}$$

$$7) M = \left[4 \frac{h_L (T_{sat} - T_c)}{h_{Lv}} \right]^{3/4} \left[\frac{\rho_L (\rho_L - \rho_v) g}{\mu_L} \right]^{1/4} \frac{L \times 4}{21} h^{7/4}$$

La puissance à évacuer $W = 300 \text{ W}$

$$W = 2 h_{Lv} M \Rightarrow h = 4 \text{ mm}$$